UV Lamp Flux Calculations and Photolysis of Planetary Ice Analogs Containing H₂O+CO (1:1) 4 = 3

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INTRODUCTION

Ultraviolet photolysis can cause major changes in the compositions of ices in the interstellar medium and on icy planetary surfaces. Our study focused on two goals: (i) the determination of the flux of the UV-emitting microwave discharge H₂ flow lamp in the UAB astrophysics lab, and (ii) the production of near-IR spectra of products formed in the photolysis of H₂O+CO (1:1) mixture at 10 K. Near-IR spectra of the photolyzed ices may be compared to observational data from ground- and space-based telescopes. We focused on near-IR spectra due to their wide use by planetary astronomers.



While UV radiation can only penetrate the top 0.1-1 μ m of an ice, the very low strengths of near-IR absorption features require 10-100 μ m of ice in order to detect them. Thus, ices had to be built up in stages (Richey 2006). We did this in the following manners:

•Simultaneous Method: A series of 30-minute deposits simultaneously exposed to the UV radiation from the lamp.

•Combination Method: 10-minute deposits with simultaneous photolysis followed by an additional 20 minutes of photolysis.

The flux of the lamp was calculated by making an ~1 µm deposit of O₂ and photolyzing the ice for a total time of 1920 seconds. The flux of the UV lamp (ϕ) is then determined from the growth of the 1040 cm⁻¹O₃ absorption feature using the equation:

$$=\frac{\Delta t \cdot \ln(10) \int \alpha dv}{2 \cdot B(O_3)}$$

where $\varDelta t$ is the photolysis time, $\alpha(v)$ is integrated over the O₃ absorption feature at 1040 cm⁻¹, and B(O₃) is the band strength of the 1040 cm⁻¹ feature (equal to 1.3 x 10⁻¹⁷ cm/molecule).



0.10

RESULTS



interior at the beginning of our experiment (0/2007), alter the inst cut-back of the deposition tube blocking the UV-Lamp (7/3/07), and after the final cut-back of the deposition tube (7/8/07).



10K mid-IR absorption spectrum of ice mixture H₂O+CO (1:1) after 62 minutes of deposition while being photolyzed, and 140 minutes of only UV-photolysis (combination method).



Column Densities of 1040 cm⁻¹ O₃ Features

 $\phi_{6/28/07}$ =4.83×10¹² photons cm² s⁻¹ $\phi_{7/0307}$ =9.43×10¹² photons cm² s⁻¹ $\phi_{7/08/07}$ =9.43×10¹² photons cm² s⁻¹

Features in the Near IR After Photolysis of H_2O+CO (1:1)



10K near-IR absorption spectrum of ice mixture H₂O+CO (1:1) after 1210 minutes of deposition while being photolysic graund 460 minutes of only photolysic simultaneous and combination methods). Also a possible leak in the system added to the amount of H₂O+CO₂ in the system overnight.



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CONCLUSIONS

In our mid-IR spectra of the photolyzed H₂O+CO ice mixture, we found features of CO₂ (at 2345 cm⁻¹) and H₂CO (1700 cm⁻¹). We can see faint features of H₂O, CO, and possibly CH4 in the near-IR. Insufficient amounts of CO₂ and H₂CO were created in order to detect them in the near-IR.

By increasing the flux of the lamp, we may yield greater amounts of ice that have strong features in the near-IR, which is the overall goal of the project. While we were able to increase the flux by a factor of 10, we are still well below the flux of 2003 $(1.1 \times 10^{14} \text{ photons cm}^{-2} \text{ s}^{-1})$.

FUTURE GOALS

Our future work involves the use of a new closed gas cell system to make the ice deposits. The cell is a three-sided capsule with a window on each side. Two of the windows are made of KBr and will allow the transfer of the spectrometer's IR beam, while the third window is made of MgF₂ in order to allow the input of UV radiation for photolysis. The UV window will be heated in order to prevent ice from forming there and blocking the UV photons.





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